

Science Learning Map

Biology

Chemistry

Physics

| Year | Autumn | Spring | | | | |
|--|---|---|---|--|--|--|
| YR7 (7 lessons over two weeks) | <u>Safety in Science</u> – Equipment and hazards. <u>Chemical Changes (pt.1)</u> – Types of reaction, word equations, metal reactions with acids, water and oxygen. <u>Cell Biology</u> – Animal and plant cells, specialised cells, unicellu organisms. <u>Energy (pt.1)</u> – Conduction, convection and radiation, insulation renewable and non-renewable resources. <u>Assessment 1</u> <u>Energy Changes</u> – endothermic and exothermic reactions. <u>Chemical Analysis</u> – Mixtures, solutions and solubility, evaporation, distillation, chromatography. <u>Space</u> – The solar system, night and day, seasons, phases of th moon. | Infection and response – Microorganisms, diseases, defence against disease. Particle model of matter – The particle model, changes of state, melting and boiling points, diffusion, pressure. Atomic structure & The Periodic Table – Metals and non-metals, The periodic Table, elements, compounds and formulae, Group 1, 7 and 0 elements. Mid-year assessment Organisation (pt.1) – The skeleton, joints, muscles, the respiratory system, smoking, drugs and alcohol, plant reproduction. Waves (pt.1) – Types of wave, sound, loudness and pitch, The Electromagnetic Spectrum, echoes and ultrasound. | <u>Assessment 2</u> <u>Organic Chemistry</u> – <u>Ecology</u> – Food chain predator/prey cycles biodiversity. <u>Atomic Structure</u> – T atom, basic types of <u>Forces (pt.1)</u> – Force forces, drag forces an <u>Investigation skills</u> – averages, anomalous | | | |
| YR8 (7 lessons over two weeks) | Homeostasis & Response – Adolescence, reproductive system The Menstrual Cycle, Fertilisation & implantation, developmer a foetus, contraception. Chemical changes (pt.2) – Acids & alkalis, the pH scale, making indicators, making salts, neutralisation. Energy (pt.2) – Calculating energy & power, cost of electricity a energy bills, energy in foods, energy stores & transfers, calcula kinetic energy and gravitational potential energy. Chemistry of the atmosphere – The atmosphere, the structure the Earth, the rock cycle. Assessment 1 Organisation (pt.2) – cells, tissues, organs and organ systems, nutrients & a balanced diet, food tests, the digestive system, enzymes & bacteria. Electricity – Static electricity, circuit symbols & drawing circuit diagrams, measuring potential difference and current, introduction to electricity formulae and rearranging formulae. | Electricity – Resistance & calculating resistance, basic calculations. Structure & properties – Properties of ionic and simple covalent substances, properties of giant covalent substances, applications of graphene. Quantitative Chemistry – Formulae, calculating relative formula mass, calculating percentage composition. Bioenergetics – Review plants cells and leaf structure, plant minerals and fertilisers, photosynthesis, limiting factors for photosynthesis, plant adaptations, aerobic respiration, anaerobic respiration. Mid-year Assessment Magnetism – Magnets and magnetic fields, electromagnets and building electromagnets, investigating electromagnets. Rate of Reaction – Basic collision theory, methods to measure rate of a reaction, investigating how concentration impact rate of reaction, catalysts. Waves (pt.2) – Types of wave, reflection, refraction, calculating speed of waves, coloured & filters, the eye. | Assessment 2 Inheritance, variation inheritance, basic genethics, variation. Using Resources – Exelectrolysis. Forces (pt.2) – Calcul motion, calculating w Investigation skills – averages, anomalous | | | |
| Year 9 will follow a new order of study agreed on by all secondary school within the MAT | | | | | | |
| YR9 (Trilogy) (9 lessons over two weeks) | <u>Cell Biology (pt.1)</u> – Animal and plant cells, Eukaryotes and prokaryotes, comparing cells, required practical for microscopy specialised cells, chromosomes, gene and DNA, mitosis and bir fission, meristems. <u>Atomic Structure and The Periodic Table (pt.1)</u> – Elements, compounds and formulae, word and chemical equations, filtration, evaporation and distillation, structure of an atom an electronic configuration, comparing models of the atom, the development of the periodic table, the modern periodic table, group 1, group 7, group 0. | Structure, bonding and the properties of matter (pt.2) – properties of simple covalent substances, polymers, giant covalent structures, fullerenes, metals and metallic bonding, alloys, states of matter and changes of state. Particle model of matter (pt.1) – Calculating density, required practical for density of regular and irregular objects. Atomic structure (pt.1) – Comparing models of the atom, Rutherford's scattering experiment, radioactive decay, isotopes, nuclear equations. | <u>Assessment 2</u> <u>Cell Biology (pt.3)</u> – I osmosis, active trans <u>Chemical Changes (p</u> Neutralisation, acids oxides, reactions bet water, displacement, <u>Electricity (pt.2)</u> – Re characteristics, AC/D in everyday appliance | | | |



Summer

- Formation of crude oil, basic alkanes and alkenes. ns and webs, ecosystems and interdependence, s, adaptation, evolution, competition and extinction,

The structure of an atom, comparing models of the radiation, research project on Chernobyl. e diagrams, calculating weight, contact & non-contact and friction, balanced & unbalanced forces.

 graphs, method writing, variables, calculating is results, conclusions, evaluations.

on & evolution – The structure of DNA, extracting DNA, enetic diagrams, cell division, genetic modification &

xtracting metals, reduction, displacement, basic

Ilating speed, motion graphs, acceleration & relative work done.

- graphs, method writing, variables, calculating s results, conclusions, evaluations.

Perfect exchange surfaces, required practical for sport.

<u>pt.1</u> – The pH scale, acids, bases and alkalis,

s and carbonates, required practical for acids and metal tween acids and metals, reactions between metals and t, reduction, electrolysis.

esistance in series, thermistors, LDR's, I/V

DC mains electricity, calculating power, energy transfers ces, I=Qt, E=QV, The National Grid.

| | 3. | Energy (pt.1) – Energy stores and systems, energy transfers, | 4. | Infection and Response (pt.1) – Pathogens, communicable | 5. | Energy Changes – Er |
|---|--|--|--|--|--|---|
| YR9 (Biology) | 3. 4. 5. 6. 7. 8. 1. | Energy (pt.1) – Energy stores and systems, energy transfers, kinetic energy, gravitational potential energy, elastic energy, Hooke's Law, specific heat capacity. Organisation (pt.1) – Cell organisation, the digestive system, nutrients, required practical for food tests, enzymes. Structure, bonding and the properties of matter (pt.1) – ionic bonding, covalent bonding, properties of ionic compounds. Assessment 1 Electricity(pt.1) – Circuit symbols and building basic circuits, measuring current and potential difference in series and parallel circuits, calculating charge and current, resistance and Ohm's Law, required practical for resistance of a wire. Organisation (pt.2) – Required practical for enzymes, the circulatory system, the heart, blood composition, the lungs and gas exchange. Cell biology (pt.1) – Plant and animal cells, using microscopes and microscopy, calculating magnification, specialised cells, prokaryotic and eukaryotic cells, comparing light and electron microscopes, Chromosomes and DNA, mitosis and binary fission, etem calls and marked parallel circuits and marked protection difference on the series of the series and parallel circuits of the series of the series of the series of the series and parallel circuits, calculating magnification, specialised cells, prokaryotic and eukaryotic cells, comparing light and electron microscopes, Chromosomes and DNA, mitosis and binary fission, etem calls and marked parallel circuits are series and parallel circuits and marked parallel circuits are series and parallel circuits are series and parallel circuits are series and parallel circuits, calculating magnification, specialised cells, prokaryotic and eukaryotic cells comparing light and electron microscopes, Chromosomes and DNA, mitosis and binary fission, etem calls are series aread parallel circuits are series are serie | 4. 5. 6. 7. 8. 1. | Infection and Response (pt.1) – Pathogens, communicable diseases, symptoms, treatments, prevention, fighting disease, white blood cells, vaccination, immunity. Bioenergetics (pt.1) – Plant cells and leaf structure, photosynthesis and starch test, limiting factors, required practical for photosynthesis and light intensity. Quantitative Chemistry – Conservation of mass, relative formula mass, percentage composition, mole calculations, reacting masses, titrations, calculating concentrations. Energy(pt.2) - transfers, systems, efficiency, calculating power, energy resources and their uses, trends in energy resources. Cell Biology (pt.2) – Stems cells, diffusion, osmosis, perfect exchange surfaces. | 5. 6. 7. 8. 1. 2. 3. | Energy Changes – Er required practical fo Organisation (pt.3) – factors, coronary hea valves, cancer. Infection and Respo drugs, drug trials, an Investigation skills – drawing graphs, mat evaluations etc. Assessment 2 Cell biology (pt.2) – as graphs and data a Organisation (pt.2) – |
| (3 lessons over two weeks) | 2. 3. | Assessment 1 Organisation (pt.1) – The digestive system, enzymes, nutrients in food and food tests, required practical for food tests. | 2. | photosynthesis and testing a leaf for starch, limiting factors, required practical and impact of light intensity on rate of photosynthesis. | | statins, artificial hear risk factors, caner. |
| YR9 (Chemistry) (3 lessons over two weeks) | 1. 2. 3. 4. | Atomic structure and The Periodic Table (pt.1) – Elements, compounds and mixtures, chemical equations and formulae, separating substances, atomic structure, electronic structure, comparing models of the atom. Bonding structure and the properties of matter (pt.1) – Formation of ions, ionic bonding, properties of ionic substances, covalent bonding, properties of simple covalent and giant covalent substances, metals and alloys, states of matter. Assessment 1 Quantitative Chemistry (pt.1) – Conservation of mass, calculating relative formula mass, calculating percentage composition, mole calculations | 1. | Chemical changes (pt.1) – Acids and alkalis, the pH scale, neutralisation, making salts, acids and carbonates, required practical for making copper sulphate, reactions of acids and metals, displacement. Energy changes – Endothermic and exothermic reactions, reaction profiles, calculating overall energy change from bod energy calculations, cell and batteries, fuel cells. | 1. 2. 3. | Assessment 2 Atomic structure an relative abundance, table, The early Perio their reactivity. Structure, bonding a properties of simple substances, Properti polymers, nanoparti |
| YR9 (Physics) (3 lessons over two weeks) | 1. 2. 3. | Energy (pt.1) – Energy stores and systems, calculating kinetic energy and gravitational potential energy, calculating specific heat capacity, required practical for specific heat capacity, Conduction, convection, insulation. Assessment 1 Electricity (pt.1) – Static electricity, electric fields, building circuits, series and parallel circuits, measuring current and potential difference, resistance and Ohm's law, required practical investigating how length of wire impact resistance. | 1. 2. 3. | Particle model – calculating density for regular and non-regular shapes, required practical for density, internal energy, calculating specific latent heat, particle motion and pressure in gasses, Brownian motion. Atomic Structure – Developing the model of the atom, atomic structure, isotopes and radiation, nuclear equations. Energy (pt.2) – Energy transfers and investigating insulation, efficiency, calculating, power, energy resources and their uses. | 1. 2. 3. | Assessment 2 Energy (pt.3) – Wind wave power, Biofuel resource use. Electricity (pt.2) – Re parallel circuits, V-I g power of electrical d |
| | | Year 10 & 11 will continue to follow the previous or | der | of study to make sure they have covered all topics a | nd a | are fully prepared |
| YR10 (Trilogy) (9 lessons over two weeks) | 1. | Structure, bonding and the properties of matter – Forming ions, ionic bonding, properties of ionic substances, covalent bonding, properties of simple covalent substances, properties of giant covalent structures, metallic bonding, alloys, polymerisation, states of matter and changes of state. Bioenergetics – Plant cells and leaf structure, photosynthesis and starch test, limiting factors, required practical for photosynthesis and light intensity, respiration and metabolism, anaerobic respiration, effect of exercise | 1. | Waves– Transverse and longitudinal waves, waves experiments and the wave equation, reflection, refraction, electromagnetic waves, properties and uses of electromagnetic waves, infra-red and temperature, required practical for black silver and white cans.Chemistry of the Atmosphere– Formation of sedimentary rocks, fossil fuels, composition of the atmosphere, evolution of the atmosphere, global warming, greenhouse effect, climate change, analysing data and graphs, impact of human activities on the | 1. 2. 3. | Assessment 2 WORK EXPERIENCE Forces – Contact and weight, mass and gra Hooke's Law, calcula graphs, required pra 1 st Law, Newton's 2 ⁿ and thinking distance |
| | | | | atmosphere, sulphur dioxide and air pollution, carbon footprint | | alkonos |

ndothermic and exothermic reactions, reaction profiles, or energy changes, bond energy calculations.

- Communicable and non-communicable diseases, risk art disease, artificial blood, artificial hearts, artificial

mse (pt.2) – fighting disease, vaccination, developing ntibiotic resistance.

- Method writing, results tables, collecting results, ths skills, averages, standard form, conclusions,

Required practical for osmosis, investigation skills such analysis, perfect exchange surfaces,

Structure of the lungs and gas exchange, heart
 on, the circulatory system, blood vessels, stents and
 onts and artificial blood, non-communicable disease and

Id the periodic table (pt.2) – Isotopes and calculating The structure and patterns in the modern Periodic odic table, group 1, 7 and 0 elements and patterns in

and the properties of matter (pt.2) – Structure and molecules, structure and properties of giant covalent ies and applications of graphene and fullerenes, icles, and their uses.

d, solar and geothermal power, hydroelectric, tides and Is and non-renewable resources, trends in energy

esistance, required practical for resistance in series and graphs, thermistors, LDR's, wiring a plug, calculating devices, The national Grid.

d for their GCSE exams

d non-contact forces, scale diagrams, resolving forces, avity, work done and power, required practical for ating speed, calculating acceleration, distance-time actical for acceleration, velocity-time graphs, Newton's rd Law, Newton's 3rd Law, stopping distances, breaking es, momentum, changes in momentum.

- Crude oil, alkanes, fractional distillation, cracking,

| | 3. 4. 5. 6. 7. | Atomic structure– Plum pudding model of the atom and Rutherford scattering, development of the nuclear model of the atom, isotopes, types of radiation, properties sand uses of each type of radiation, half-life, nuclear equations, background radiation.Quantitative Chemistry– Relative formula mass, percentage composition, mole calculations, reacting masses, titrations, concentration calculations.Assessment 1 Homeostasis– The nervous system, Synapses and reflexes, required practical for reaction times, the endocrine system, controlling blood glucose levels, diabetes, adrenaline and thyroxin, puberty, menstrual cycle, fertility hormones, using fertility hormones, contraception.Energy Changes– Endothermic and exothermic reactions, reaction profiles, required practical for energy changes, bond | 3. | Particle Model – Density and calculating density of regular and irregular objects, motion in gasses, gas pressure, internal energy, changes of state, specific latent heat. Inheritance, Variation and Evolution – Sexual/Asexual reproduction, chromosomes, DNA, proteins and genomes, Meiosis, genetic diagrams, inherited disorders, mutation, variation, antibiotic resistance, evolution, fossils, selective breeding, genetic engineering, classification. | 5. | Investigation skills – drawing graphs, mat evaluations etc. |
|--|--|--|----------------|--|------------------------------|---|
| YR10 (Biology) | 1. 2. 3. | Homeostasis and response (pt.1) – The nervous system, synapses and reflexes, requires practical for reaction times, the brain, the eye, the endocrine system, controlling body temperature, controlling blood glucose concentration, diabetes, controlling blood water concentration. Assessment 1 Homeostasis and response (pt.2) – Kidney failure and dialysis, plant hormones and auxins, puberty, the menstrual cycle, fertility hormones, using fertility hormones. | 1. | Homeostasis and response (pt.3) – Contraception, plant hormones, commercial uses for plant hormones. Inheritance, variation and evolution – DNA, protein synthesis, mutations, mitosis and meiosis, sexual and asexual reproduction, genetic diagrams, inheriting disorders, variation, evolution and natural selection, Darwin and Lamarck, extinction, selective breeding, genetic engineering, cloning. | 1. 2. 3. | Assessment 2 WORK EXPERIENCE Ecology – Competitio adaptations, extrem the water cycle, the management, global trophic levels, pyram biotechnology. |
| YR10 (Chemistry) | 1. 2. 3. | Rate of reaction – Collision theory, required practical concentration, surface area, temperature, catalysts, calculating rate of reaction from graphs, tangents, reversible reactions, Le Chateliers Principle. Assessment 1 Chemical analysis (pt.1) – Purity and formulations, chromatography and Rf values, testing for gases. | 1. | <u>Chemical analysis (pt.2)</u> – testing for cations flame tests, testing for cations precipitation, testing for anions carbonates, testing for anions sulphates, testing for anions halides, required practical testing for unknown compounds s=using chemical tests, flame emission spectroscopy. <u>Organic chemistry</u> – Crude oil and alkanes, fractional distillation, cracking, alkenes, testing for saturation, addition polymerisation, hydrogenation, making ethanol ethane and fermentation, alcohols, carboxylic acids, esters. | 1. 2. 3. | Assessment 2 WORK EXPERIENCE Chemistry of the atm of the atmosphere, g impact of human act the greenhouse effe |
| YR10 (Physics) | 1. 2. 3. 4. | <u>Space physics</u> – The universe and the solar system, lifecycle of stars, orbits and satellites, The Big Bang Theory, Evidence for The Big Bang Theory. <u>Forces (pt.1)</u> – Contact and non-contact forces, weight, mass and gravity, centre of mass, force diagrams, resolving forces, scaled balanced forces, work done, Hooke's Law. <u>Assessment 1</u> <u>Forces (pt.2</u>) – Required practical for Hooke's Law, moments, fluid pressure, distance, displacement, speed and velocity. | 1. | Forces (pt.3) – distance-time graphs, acceleration, velocity-time graphs, required practical for acceleration, terminal velocity, Newton's 1 st Law, Newton's 2 nd Law, stopping distances, thinking and breaking distances, momentum, change in momentum, car safety and momentum. | 2. 3. 4. | Assessment 2 WORK EXPERIENCE Magnetism and elect electromagnetism the theory, building an e microphones, transf |
| YR11 (Trilogy) (8 lessons over two weeks) | 1. 2. 3. 4. | Rate of Reaction – collision theory, required practical investigating concentration, temperature, surface area, catalysts, calculating rate of reaction from a graph, tangents and gradients, endothermic and exothermic reactions, Le Chatelier's Principle. Magnetism and Electromagnets – Permanent and induced magnets, electromagnets, investigating electromagnets, right hand rule, the motor effect, left hand rule, electric motors, applications of the motor effect. Using Resources – Finite and renewable resources, reuse and recycle, life cycle assessments, potable water and waste water treatments, required practical purification, required practical finding mass of a solute. Assessment 1 | 1. 2. 3. | Inheritance, Variation and Evolution – Sexual/Asexual reproduction, chromosomes, DNA, proteins and genomes, Meiosis, genetic diagrams, inherited disorders, mutation, variation, antibiotic resistance, evolution, fossils, selective breeding, genetic engineering, classification. <u>Assessment 2</u> <u>Revision</u> - Lessons will be revision based until the GCSE exams. These lessons will be determined from analysis of both sets of mock exams using questions level analysis (QLA's) to inform teachers and students of topics and skills that need extra attention during lessons and intervention. | us | Lessons wi These lessons will be ing questions level an and skills that nee |

- Method writing, results tables, collecting results, ths skills, averages, standard form, conclusions,

ion, food chains and webs, abiotic and biotic factors, nophiles, collecting ecological data, quadrats, transects, e carbon cycle, decay, biodiversity and waste al warming, deforestation, land use and peat bogs, mids of biomass, food security and farming, fish stocks,

mosphere – Composition of the atmosphere, evolution global warming, data analysis from tables and graphs, tivities on the atmosphere, peat bogs, carbon footprint, ect, plastics and recycling.

<u>ctromagnets</u> – Permanent and induced magnets, heory, investigating electromagnets, the motor effect electric motor, loudspeakers, the generator effect, formers.

TBD

vill be revision based until the GCSE exams. determined from analysis of both sets of mock exams nalysis (QLA's) to inform teachers and students of topics ed extra attention during lessons and intervention.

| | 5. | Forces – Contact and non-contact forces, scale diagrams, resolving forces, weight, mass and gravity, work done and power, required practical for Hooke's Law, calculating speed, calculating acceleration, distance-time graphs, required practical for acceleration, velocity-time graphs, Newton's 1 st Law, Newton's 2 nd Law, Newton's 3 rd Law, stopping distances, breaking and thinking distances, momentum, changes in momentum. | | | |
|---------------------|-----------------------------------|---|---|---|---|
| YR11 (Biology) | 1. 2. 3. 4. | <u>Cell biology</u> – Movement in and out of cells, mitosis and binary fission, stem cells. <u>Organisation</u> – Required practical for enzymes, cardiovascular system, Required practical for food tests, transpiration and translocation. <u>Assessment 1</u> <u>Infection and Response</u> – Diseases, fighting diseases, developing drugs, monoclonal antibodies, plant defences. | 1. 2. 3. 4. | <u>Homeostasis and Response</u> – The nervous system, the brain, the eye, the endocrine system, controlling temperature, blood glucose, blood water concentration, fertility. <u>Inheritance, Variation & Evolution</u> – DNA structure, mutations, reproduction, meiosis, genetic diagrams, evolution, selective breeding, genetic modification, cloning. <u>Assessment 2</u> <u>Ecology</u> – Required practical for quadrats, line transects, explaining ecosystems, required practical for decay, the water cycle and the carbon cycle, maintaining biodiversity, trophic levels and pyramids of biomass, food security and biotechnology. | Lessons w These lessons will be using questions level an and skills that nee |
| YR11 (Chemistry) | 1. 2. 3. 4. | Quantitative chemistry – Relative formula mass, percentage composition, moles and reacting masses, yield and atom economy, gases and solutions, titration, concentration calculations. Assessment 1 Chemical changes – Required practical for making salts, electrolysis. Using Resources – Potable water, waste water, lifecycles. | 1. 2. 3. 4. 5. | Structure, bonding and properties of matter – lonic bonding, covalent bonding, comparing structures of diamond, graphite, graphene, properties of ionic substances and simple covalent substances.Chemical changes – Electrolysis, redox reactions.Assessment 2Rate of Reaction – Equilibrium, The Haber Process.Organic Chemistry – Alkanes, fractional distillation, alkenes, polymers. | Lessons w These lessons will be using questions level an and skills that nee |
| YR11 (Physics) | 1. 2. 3. 4. 5. | Atomic structure– Modern theory of the atom, development of the model of the atom, types of radiation, uses of radiation, nuclear equations, half-life, background radiation and contamination, Chernobyl and nuclear power, nuclear fission, nuclear fusion.Energy– Energy stores and transfers, conduction, convection, insulation, required practical for specific heat capacity, energy sources.Assessment 1Electricity – Series and parallel circuits, required practical for investigating resistance, electricity equations, static electricity and electric fields.Particle model of matter energy, specific latent heat calculations, gas motion and pressure. | 1. 2. 3. 4. 5. 6. | Atomic structure – types of radiation, structure of the atom, nuclear equations, half-life, contamination, nuclear fission, nuclear fusion. Forces – Contact and non-contact forces, weight, mass and gravity, force diagrams, Hooke's Law, fluid pressure, moments, motion graphs, Newton's Laws, stopping distances, momentum. Space Physics – Lifecycle of stars, orbits and satellites, evidence for The Big Bang Theory. Assessment 2 Waves – Types of wave and properties, EM waves, lenses. Magnetism and Electromagnetism – Magnets and electromagnets, the motor effect, Electromagnetic induction, transformers. | Lessons w These lessons will be using questions level an and skills that nee |

Aspire for Excellence

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vill be revision based until the GCSE exams. determined from analysis of both sets of mock exams nalysis (QLA's) to inform teachers and students of topics ed extra attention during lessons and intervention.

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